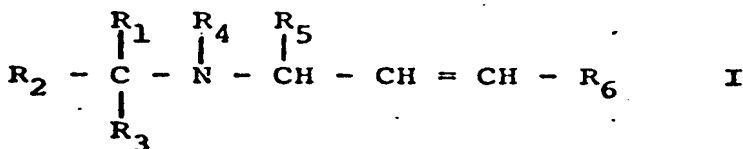


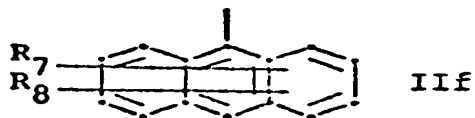
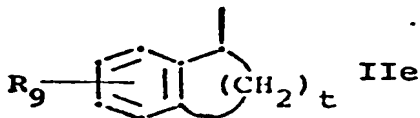
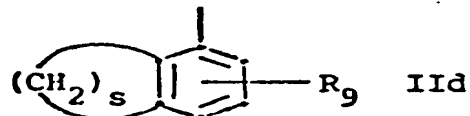
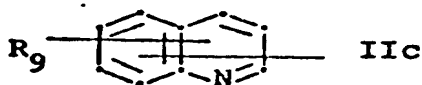
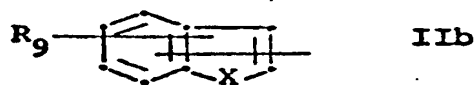
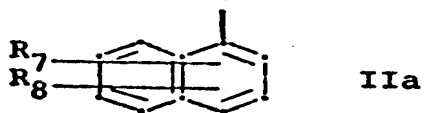
PROPENYLAMINES, PROCESSES FOR THEIR PRODUCTION,  
PHARMACEUTICAL COMPOSITIONS CONTAINING THEM AND THEIR  
USE AS PHARMACEUTICALS

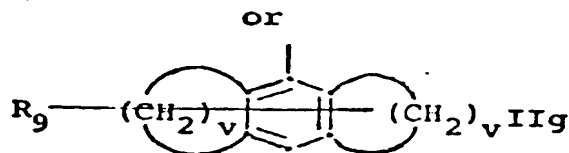
SV P This invention relates to propenylamines,  
 processes for their production, pharmaceutical compositions  
 containing them and their use as pharmaceuticals.

The invention provides compounds of formula I,



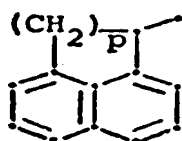
5 ps wherein (a)  $R_1$  represents a group of formula



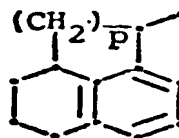


<sup>ps</sup> and  $R_2$  represents hydrogen or lower alkyl,  
or  $R_1$  and  $R_2$  together represent a group of formula

T0040X



IIh



IIIi

<sup>ps</sup> whereby in the formulae IIa to IIIi,

<sup>po</sup>  $R_7$  and  $R_8$  represent, independently, hydrogen, halogen, tri-  
5 fluoromethyl, hydroxy, nitro, lower alkyl or lower alkoxy,

<sup>po</sup>  $R_9$  represents hydrogen, halogen, hydroxy, lower alkyl or  
lower alkoxy,

<sup>po</sup> X represents oxygen, sulphur, imino, lower alkyl imino or  
a radical of formula  $-(CH_2)_r-$ ,  
13 13

10 <sup>po</sup> p is 1, 2 or 3,

r is 1, 2 or 3,

s is 3, 4 or 5,

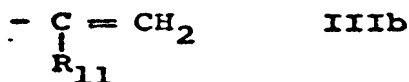
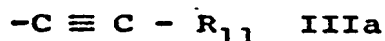
t is 2, 3 or 4, and

v is 3, 4, 5 or 6;

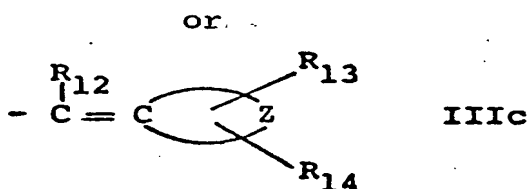
15  $R_3$  and  $R_5$  represent, independently, hydrogen or lower alkyl,  
and

<sup>po</sup>  $R_4$  represents  $C_{1-6}$  alkyl or  $C_{3-8}$  cycloalkyl- $(C_{1-6})$ -alkyl;  
and

<sup>po</sup>  $R_6$  represents a group of formula



T0050X



$P_0$  wherein  $R_{11}$  represents hydrogen, optionally  $\alpha$ -hydroxy substituted alkyl; alkenyl, alkynyl, cycloalkyl, cycloalkylalkyl, phenyl, phenalkyl or thienyl,

5  $P_1$   $R_{12}$ ,  $R_{13}$  and  $R_{14}$  represent, independently, hydrogen or lower alkyl, and

T0051X



$P_0$  represents a  $C_{5-8}$  cycloalkylidene radical optionally containing a double bond; or

10  $P_0$  (b)  $R_1$  represents a group of formula IIa to IIg as defined under (a),

$P_0$   $R_2$  represents hydrogen or lower alkyl,

$R_3$  and  $R_4$  together form a group  $-(CH_2)_u-$ , wherein  $u$  is an integer of 1 to 8, and

$P_0$   $R_5$  and  $R_6$  have the meanings given under (a).


15  $P$  Any lower alkyl or lower alkoxy radical has preferably 1 to 4 carbon atoms, especially 2 or 1 carbon atoms. Unless otherwise stated alkyl moieties preferably have 1 to 12 carbon atoms especially 2 to 8 carbon atoms,

particularly 2 to 6 carbon atoms and most preferably 3 to 5 carbon atoms and if bridging 1 to 4 particularly 1 or 2 carbon atoms. Any alkenyl or alkynyl radical has preferably 3 to 6 carbon atoms, especially 3 or 4 carbon atoms, e.g. allyl, propenyl or propynyl. Such alkyl, alkoxy, alkenyl and alkynyl groups can be straight-chain or branched. A preferred cycloalkylidene radical is cyclohexylidene. The term cycloalkyl is to be understood as including polycyclo groups such as bornyl or adamantyl but is preferably cyclohexyl or cyclopentyl.

Conveniently  $R_7$  and  $R_8$  are identical and are both hydrogen. Conveniently  $R_9$  is hydrogen or halogen. In IIb and IIc the bond to the carbon atom to which  $R_2$  and  $R_3$  are attached is conveniently attached meta to X and para to the ring nitrogen, respectively. X is conveniently sulphur, imino or lower alkylamino.  $R_1$  is preferably a radical of formula IIb, IIc or IIId, or especially IIa.  $R_2$  is preferably hydrogen.  $R_3$  is preferably hydrogen and  $R_4$  is conveniently alkyl.  $R_5$  is conveniently hydrogen.

The values of p, r, s, t, u and v are conveniently chosen to produce a seven- preferably a five- or six-membered ring.

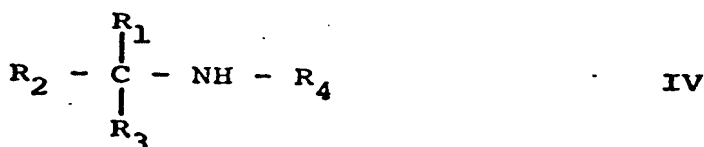
The double bond between  $R_6$  and the nitrogen atom preferably has the trans-configuration.



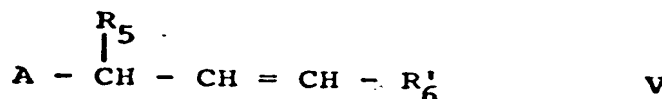
Halogen stands for fluorine, chlorine or bromine, preferably chlorine or bromine.

The present invention also provides a process  
5 for the production of a compound of formula I, which comprises

P (a) when  $R_6$  represents a group of formula IIIa, as defined  
above, (compound Ia), reacting a compound of formula IV,

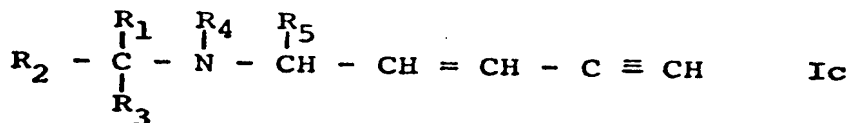


PS wherein  $R_1$  to  $R_4$  are as defined above, with a com-  
10 pound of formula V,



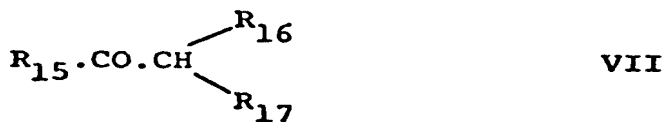
PS wherein A is a leaving group,  $R_5$  is as defined above,  
and  $R'_6$  stands for a group of formula IIIa, as  
40 defined above, or

P (b) when  $R_6$  represents a group of formula IIIa, wherein  $R_{11}$   
15 represents  $\alpha$ -hydroxyalkyl (compounds Ib), reacting a metal-  
ated compound of formula Ic,



PS wherein  $R_1$  to  $R_5$  are as defined above, with a carbonyl compound of formula VII,

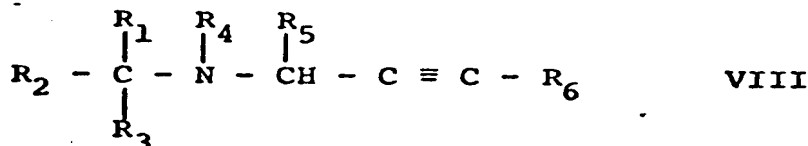
T0080X



wherein  $R_{15}$ ,  $R_{16}$  and  $R_{17}$  represent independently hydrogen or lower alkyl, or

5P(c) when the double bond between  $R_6$  and the nitrogen atom is in trans configuration (compounds Id) reducing a compound of formula VIII,

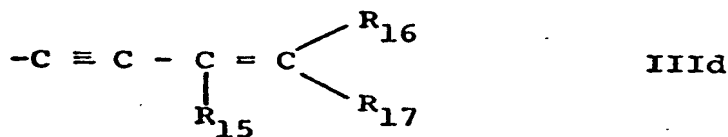
T0081X



PS wherein  $R_1$  to  $R_6$  are as defined above, with diisobutyl-aluminiumhydride, or

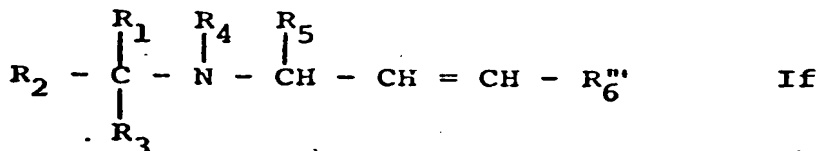
10P(d) when  $R_6$  represents a group of IIIb or IIIc as defined above or a group of formula IIId,

T0082X



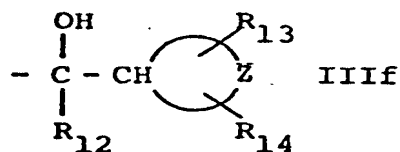
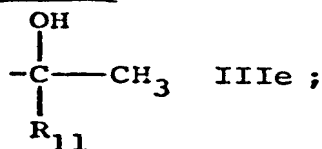
PS wherein  $R_{15}$ ,  $R_{16}$  and  $R_{17}$  are as defined above (compounds Ie) splitting off water from a compound of formula

T0083X

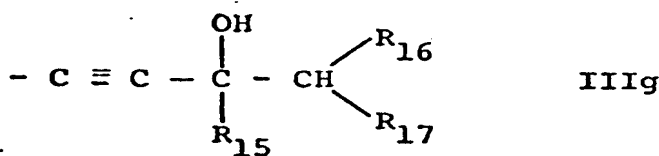


8

PS wherein  $R_1$  to  $R_5$  are as defined above,  
 and  $R_6$  represents a group of formula IIIe, IIIf,  
 or IIIg,

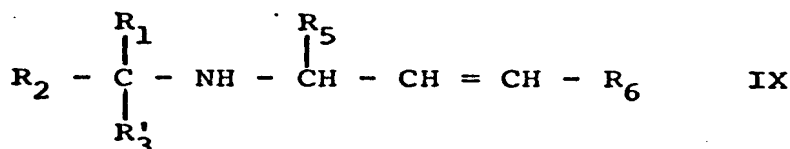


T0090X



PS wherein  $R_{11}$  to  $R_{17}$  and Z are as defined above, or  
 5 P(e) when  $R_3$  represents hydrogen or lower alkyl and  $R_4$   
 represents  $C_{1-6}$ alkyl or  $C_{3-8}$ cycloalkyl- $(C_{1-6})$ -alkyl (com-  
 pounds Ig), introducing the group  $R_4$  into a compound of  
 formula IX,

T0091X



PS wherein  $R_1$ ,  $R_2$ ,  $R_5$  and  $R_6$  are as defined above,  
 10  $R_3$  represents hydrogen or lower alkyl, and  
 $R_4$  represents  $C_{1-6}$ alkyl or  $C_{3-8}$ cycloalkyl-  
 $(C_{1-6})$ -alkyl.

P Process (a) may be effected in conventional  
 manner for the production of tertiary amines by conden-  
 15 sation from analogous starting materials. The process may

9

be effected in an inert solvent such as a lower alkanol, e.g. ethanol, optionally in aqueous admixture, an aromatic hydrocarbon solvent, e.g. benzene or toluene, a cyclic ether, e.g. dioxane or a carboxylic acid dialkylamide solvent, e.g. dimethylformamide. The reaction temperature is conveniently from room temperature to the boiling temperature of the reaction mixture, preferably room temperature. The reaction is conveniently effected in the presence of an acid binding agent, such as an alkali metal carbonate, e.g. sodium carbonate. The leaving group A is conveniently iodine or preferably chlorine or bromine, or an organic sulphonyloxy group having 1 to 10 carbon atoms, e.g. alkylsulphonyloxy, preferably having 1 to 4 carbon atoms such as mesyloxy, or alkylphenylsulphonyloxy preferably having 7 to 10 carbon atoms such as tosyloxy.

Process (b) may be effected in conventional manner, for example by metalating the compound of formula Ic, e.g. with butyllithium in an inert solvent such as an ether e.g. tetrahydrofuran and subsequently reacting the metalated compound of formula Ic, thus obtained, preferably without isolation with a compound of formula VII.

The reduction with diisobutylaluminium hydride (DIBAH) according to process (c) is preferably carried out in an inert solvent e.g. in an aromatic hydrocarbon such as toluene or benzene and at room temperature or raised temperature e.g. 35 to 40°C.

20



The splitting-off of water according to process d) can be carried out with a suitable agent such as an inorganic acid, e.g. hydrochloric or sulphuric acid, an organic acid, e.g. methanesulphonic acid, benzenesulphonic acid or p-toluenesulphonic acid or an inorganic or organic acid anhydride or -halide e.g.  $\text{POCl}_3$  in an inert solvent. An excess of an acid halide if used can act as reaction medium whereby the reaction is carried out in the presence of an acid binding agent such as a tertiary amine, e.g. a trialkylamine or ~~pyridine~~ <sup>3/</sup>pyridine. Reaction temperatures vary according to reaction conditions and lie for example between -10 and 180°C. The splitting-off of water can also be carried out with the help of polyphosphoric acid at temperatures between 80 and 120°C whereby inorganic acids such as phosphoric acid, organic acids such as acetic acid or an excess of polyphosphoric acid can serve as solvent.

Process (e) may be effected in manner conventional for the "alkylation" of secondary amines (the term "alkylation" being used here to denote introduction of any of the hydrocarbyl groups  $\text{R}_4$ ), for example by direct "alkylation" with an "alkylating" agent, for example a halide or sulphate, or by reductive alkylation, in particular by reaction with an appropriate aldehyde and subsequent or simultaneous reduction. Reductive "alkylation"

is suitably effected by reacting a compound of formula IX in an inert organic solvent, such as a lower alkanol, e.g. methanol, and at an elevated temperature, in particular at the boiling temperature of the reaction mixture with the corresponding aldehyde. The subsequent reduction may be effected with, for example, a complex metal hydride reducing agent, e.g.  $\text{NaBH}_4$  or  $\text{NaCNBH}_3$ . The reduction may also be effected simultaneously to the alkylation, for example by use of formic acid which may serve both as reducing agent and as a reaction medium. The reaction is preferably carried out at raised temperature, in particular at the boiling point of the reaction mixture.

Free base forms of the compounds of formula I may be converted into salt forms and vice versa. Suitable acid addition salts are e.g. hydrochloride, hydrogen fumarate or naphthalene-1,5-disulphonate.

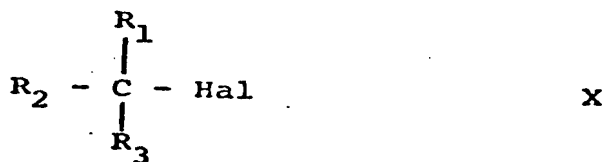
The compounds of the formula I and their intermediates can be obtained in the form of isomeric mixtures of the various cis/trans isomers which can be separated according to established methods. Alternatively, isomers of the compounds can be obtained by using the appropriate isomer of the starting material. Unless otherwise stated the compounds are always to be understood as being mixtures of these isomers.

The starting materials of formula IV are in part new and can be prepared by reacting in conventional

12

manner a compound of formula X,

T0130X



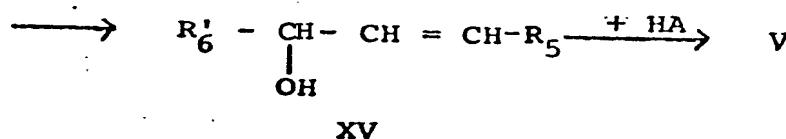
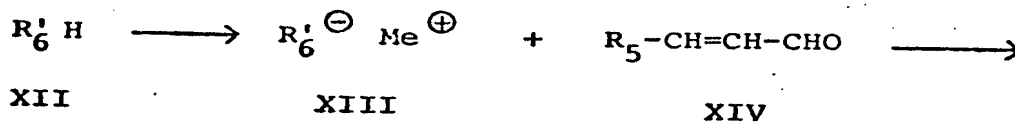
PS with a compound of formula XI, PS



PS wherein in the formulae X and XI  $R_1$  to  $R_4$  are as defined above and Hal stands for halogen.

P The starting materials of formula V are in part new and can be prepared by reacting a compound of formula XII, PS

PS according to the following scheme

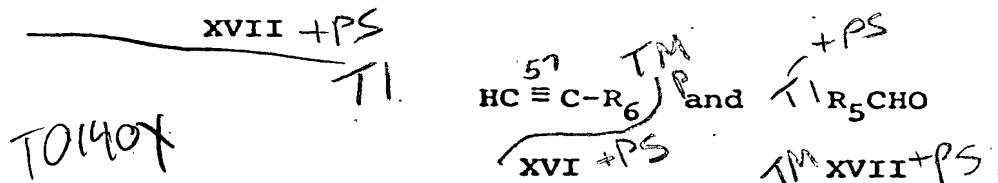


PS whereby  $R_6'$ ,  $R_5$  (and A are as defined above and  $Me \oplus$  represents a metal cation.

T0131X

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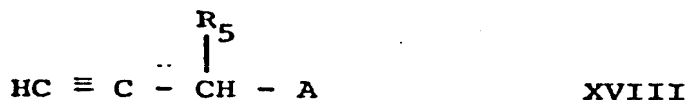
The starting materials of formula VIII are new and can be prepared (a<sub>40</sub><sup>i</sup>) by subjecting a compound of formula IV, defined above, and compounds of formulae XVI and



PS 5 to a Mannich reaction or

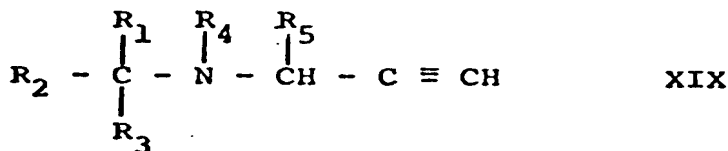
P (b<sub>40</sub><sup>i</sup>) in the case when R<sub>6</sub> represents a group of formula IIIa as defined above by reacting a compound of formula IV as defined above with a compound of formula XVIII

TO141X



PS to give a compound of formula XIX,

TO142X

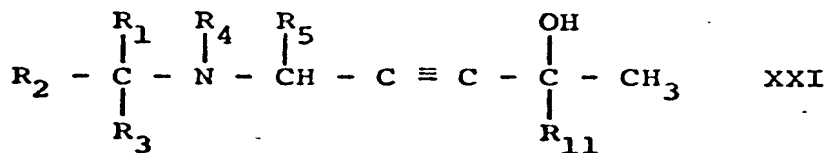


PS 10 and subjecting this to a Cadiot-Chodkiewicz coupling reaction with Cu<sub>30</sub><sup>+</sup> and a compound of formula XX, PS



PS or (c<sub>40</sub><sup>i</sup>) when R<sub>6</sub> represents a group of formula IIb as defined above splitting off water from a compound of formula XXI,

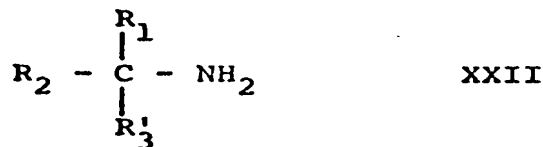
TO150X



PS whereby in the formulae XVI to XXI  $R_1$  to  $R_6$ ,  $R_6'$ ,  $R_{11}$ ,  
A and Hal are as defined above. 40

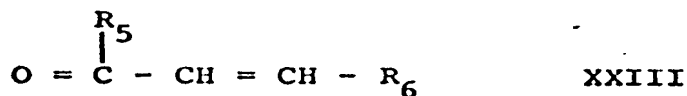
⌘ The starting materials of formula IX are new  
and can be prepared for example by reacting a compound of  
5 formula XXII,

TO151X



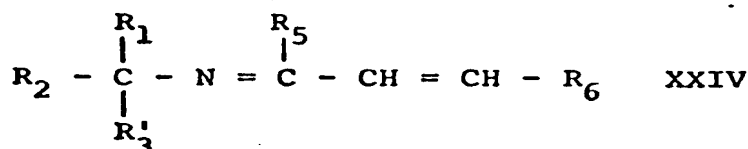
PS with a compound of formula XXIII

TO152X



PS to give a compound of formula XXIV

TO153X

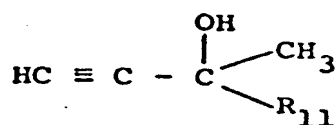


PS and reducing this e.g. with a complex hydride such as  
 $NaBH_4$ , whereby in the formulae XXII to XXIV  $R_1$ ,  $R_2$ ,  $R_3'$ ,  $R_5$   
10 and  $R_6$  are as defined above. 40

⌘ Compounds of formula XXI can be prepared  
(a") by subjecting a compound of formula IV as defined above,  
a compound of formula XVII as defined above, and a compound  
of formula XXV,

15

T0160X



XXV

PS to a Mannich reaction, or

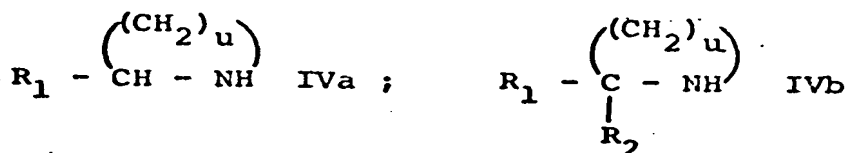
P (b<sub>u</sub>) metalating a compound of formula XIX, as defined above, and reacting the metal compound thus obtained with a carbonyl compound of formula XXVI, PS



5 PS whereby in the formulae XXV and XXVI R<sub>11</sub> is as defined above.

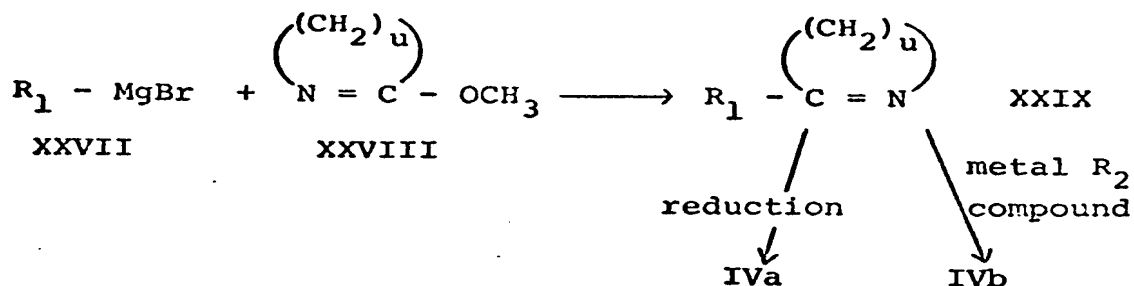
P The compounds of formulae IVa and IVb

T0161X



PS can be prepared according to the following scheme

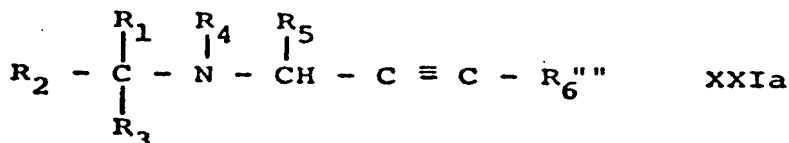
T0162X



PS whereby in the formulae IVa, IVb and XXVII to XXIX R<sub>1</sub>, R<sub>2</sub> and u are as defined above.

16

*P* The starting materials of formula If wherein <sup>4140</sup>R<sub>6</sub><sup>'''</sup> represents a group of formula IIIe or IIIf as defined above are new and can be prepared by reduction with LiAlH<sub>4</sub> of a compound of formula XXIa,



*T O I P X*

*PS* 5 wherein R<sub>1</sub> to R<sub>5</sub> are as defined above and R<sub>6</sub><sup>'''</sup> represents a group of formula IIIe or IIIf as defined above.

*P* Compounds of formula XX are in part new and can be prepared by reacting a compound of formula XII, as defined above, with butyllithium and a halogen.

10 The new compounds of formulae IV, V, VIII, IX XX and If also form part of the invention. The remaining intermediate compounds are either known or can be prepared according to known methods or as hereinbefore described.

15 The compounds of formula I are useful because they possess chemotherapeutic activity. In particular, they are useful as antimycotic agents, as indicated in vitro in various families and types of mycetes, including Trichophyton spp, Aspergillus spp, Microsporum spp and Sporotrichium schenkii and Candida spp at concentrations of, for example 0.01 to 100 <sup>82</sup>µg/ml, and in vivo in the experimental skin mycosis model in guinea pigs. In this model, guinea pigs are infected by subcutaneous applications of Trichophyton Quinckeanum. The test substance is administered

*17*

daily for 7 days beginning 24 hours after the infection either by local application by rubbing the test substance (taken up in polyethylene glycol) on the skin surface, or perorally or sub-cutaneously, the test substance being  
5 administered as a suspension. The activity is shown on local application at concentrations of for example 0.01 to 5%. The oral activity is shown in vivo in the guinea pig Trichophytosis model at dosages of, for example 2 to  
13 70 mg/kg.

10 For the above-mentioned use, the dose administered will of course vary depending on the compound employed, mode of administration and treatment desired. However, in general, satisfactory results are obtained when administered at a daily dosage of from 1 to 100 mg/kg of animal body  
15 weight, conveniently given in divided doses two to four times daily, or in sustained release form. For the larger mammals, the corresponding daily dosages are in the range of from 70 to 2000 mg, and dosage forms suitable for oral administration comprise from 17.5 to 1000 mg. The invention  
20 therefore also concerns a method of treating diseases or infections caused by mycetes using a compound of formula I.

The compounds may be used in free base form or in the form of chemotherapeutically acceptable acid addition salts. Such salt forms exhibit the same order of activity as the free base forms. Suitable salt forms are e.g.  
25 hydrochloride, hydrogen fumarate or naphthalene-1,5-disulphonate.

18



The compounds may be admixed with conventional chemotherapeutically acceptable diluents and carriers, and, optionally, other excipients and administered in such forms as tablets or capsules. The compounds may alternatively be administered topically in such conventional forms as ointments or creams or parenterally. The concentrations of the active substance will of course vary depending on the compound employed, the treatment desired and the nature of the form etc. In general, however, satisfactory results are obtained e.g. in topical application forms at concentrations of from 0.05 to 5, in particular 0.1 to 1 wt %.

Such compositions also form part of the invention.

Examples of preferred compound groups are

- 15<sup>0</sup> (i) compounds of formula I wherein  $R_6$  represents a group of formula IIIa wherein  $R_{11}$  represents alkyl preferably  $C_{2-8}$  alkyl, more preferably  $C_{2-6}$  alkyl, most preferably  $C_{3-5}$  alkyl for example n- or in particular t-butyl;

<sup>P</sup> (ii) compounds of formula I wherein  $R_6$  represents a group of formula IIIa wherein  $R_{11}$  represents  
<sup>60</sup>  $\alpha$ -hydroxy substituted alkyl; alkenyl, alkynyl, cycloalkyl, cycloalkylalkyl, phenyl, phenalkyl  
<sup>5</sup> or thienyl;

<sup>P</sup> (iii) compounds of formula I wherein  $R_{11}$  represents alkyl, alkenyl, alkynyl, cycloalkylalkyl, phenyl or phenalkyl and all other substituents are as defined under formula I;

<sup>10</sup> <sup>P</sup> (iv) compounds of formula I wherein

<sup>PS</sup> (a) <sup>P</sup>  $R_1$  represents a group of the formula IIa, IIb, IIc,

$R_2$  represents hydrogen,

$R_3$  represents hydrogen,

$R_4$  represents lower alkyl,

<sup>15</sup>  $R_5$  represents hydrogen or lower alkyl, <sup>C</sup>

or <sup>P</sup>  $R_3$  and  $R_4$  together form a group  $-(CH_2)_{13}-$  or  $-(CH_2)_{13}-$

20

PS (b) wherein <sup>P</sup>R<sub>1</sub> and R<sub>2</sub> together represent a group of the formula IIh,

5 <sup>P</sup>  
 $\left\{ \begin{array}{l} R_3 \text{ represents hydrogen,} \\ R_4 \text{ represents lower alkyl,} \\ R_5 \text{ represents lower alkyl and} \\ R_6 \text{ is as hereinbefore defined,} \end{array} \right.$

PS whereby within these groups R<sub>6</sub> is preferably a group of formula IIIa as hereinbefore defined or as described under (i) or (ii) above and/or R<sub>1</sub> is preferably a group  
 10 of formula IIa.

<sup>P</sup> Preferred meanings of the substituents in the compounds of the formula I are such as set out hereinbefore.

Compounds of formula I are generally preferred  
 15 wherein the double bond between R<sub>6</sub> and the nitrogen atom is in trans-configuration.

Particularly preferred individual compounds are:  
 N-methyl-N-(1-naphthylmethyl)-non-2(trans)-en-4-ynyl-1-amine and N-methyl-N-(1-naphthylmethyl)-6,6-dimethyl-hept-  
 20 2(trans)-en-4-ynyl-1-amine, and their hydrochlorides.

DE The following Examples illustrate the invention whereby all temperatures are in degrees centigrade.

CL  
CL  
EXAMPLE 1

trans-N-(3-Benzo[b]thiophenemethyl)-N-methyl-non-2-en-4-ynyl-1-amine and cis-N-(3-Benzo[b]thiophenemethyl)-N-methyl-non-2-en-4-ynyl-1-amine [process a)]

- 5        P        12 g 1-Bromo-2-nonen-4-yne (cis/trans mixture) are added dropwise to a mixture of 10.5 g N-(3-Benzo[b]thiophenemethyl)-N-methylamine, 8.2 g K<sub>2</sub>CO<sub>3</sub> and 100 ml dimethylformamide and stirred overnight. The reaction mixture is filtered and the solvent removed under vacuum.
- 10 The residue is partitioned between ether and saturated aqueous NaHCO<sub>3</sub>, the organic phase dried, concentrated under vacuum and chromatographed over kieselgel using toluene/ethylacetate 4:1 as eluant. The trans isomer is eluted first followed by the cis isomer. Both are oils.

CL  
L  
15 EXAMPLE 2

trans-N-Methyl-N-(1-naphthylmethyl)-6-hydroxy-6-methyl-hept-2-en-4-ynyl-1-amine [process b)]

- P        10.7 ml of a 15% butyllithium solution in hexane are added dropwise to 3g of trans N-methyl-N-(1-naphthylmethyl)pent-2-en-4-ynyl-1-amine in absolute tetrahydrofuran and reacted after 30 minutes with a solution of 1.79 g of acetone. The reaction mixture is stirred for 24 hours at room temperature, poured onto ice and extracted

22

with chloroform. The organic phase is washed, dried and concentrated under vacuum. After chromatography over kieselgel (eluant toluene/ethyl acetate 4:1) the title compound is obtained as an oil.

CL 5 EXAMPLE 3:

(a) trans-N-Methyl-N-(1-naphthylmethyl)-non-2-en-4-ynyl-1-amine [process c)]

P 72 ml of a 1.2M solution of DIBAL in toluene are added dropwise to a solution of 5g N-methyl-N-(1-naphthylmethyl)-2,4-nonadiynyl-1-amine in dry toluene and the resulting mixture stirred under protective gas overnight at 40° and then for 24 hours at room temperature.

20. The excess reagent is broken down with 2N NaOH under cooling and the reaction mixture extracted with ether.  
15 The organic phase is dried, concentrated under vacuum and chromatographed over kieselgel (eluant - toluene/ethyl-acetate 95:5). The title substance is isolated as an oil.

CL (b) Hydrochloride salt

P The compound from a) is converted to its hydrochloride in conventional manner e.g. by treating with 20 4N ethanolic HCl and melts after recrystallisation at 118-121°C.

14 20

23

CL  
L  
EXAMPLE 4

N-Methyl-N-(1-naphthylmethyl)-deca-2(trans),6(cis)-dien-4-ynyl-1-amine

- 5 1g trans-N-Methyl-N-(1-naphthylmethyl)-6-hydroxy-dec-2-en-4-ynyl-1-amine are refluxed under a water separator with 570 mg p-toluenesulphonic acid (monohydrate) in benzene. The mixture is cooled after 2 hours, the organic phase shaken a number of times with saturated aqueous NaHCO<sub>3</sub>, dried and concentrated under vacuum. The residue is chromatographed over kieselgel (eluant 10 toluene/ethylacetate 9:1) to give the title product.

CL  
L  
EXAMPLE 5

N-Methyl-N-(1-naphthylmethyl)-4-cyclohexyl-2-(trans)-4-pentadienyl-1-amine (A) and N-Methyl-N-(1-naphthylmethyl)-15 4-cyclohexylidenyl-2-(trans)-pentenyl-1-amine (B)

- 20 1g N-Methyl-N-(1-naphthylmethyl)-4-hydroxy-4-cyclohexyl-2-pentenyl-1-amine is refluxed under a water separator with 570 mg p-toluenesulphonic acid (monohydrate) in benzene. The mixture is cooled after 2 hours, the organic phase shaken a number of times with saturated aqueous NaHCO<sub>3</sub>, dried and concentrated under vacuum. The residue is chromatographed over kieselgel (eluant 20 toluene/ethyl acetate 9:1) to obtain first title product (A) followed by title product (B) as oils.

24

CL  
L  
EXAMPLE 6

trans-N-Methyl-N-(1-naphthylmethyl)-4-cyclohexylidenyl  
2-butenyl-amine [process e)]

5 3g (1-Naphthylmethyl)amine and 2.86 g  
4-cyclohexylidenyl-2-butenal are stirred in ether  
together with a 4 Å molecular sieve. The reaction mix-  
ture is filtered and concentrated under vacuum. The  
residue is taken up in methanol, treated with 800 mg  
NaBH<sub>4</sub> and stirred for 2 hours at room temperature.






10 The reaction mixture containing the secondary  
amine thus obtained is taken directly for reductive  
methylation. 8 ml 37% aqueous formaldehyde solution are  
added and refluxed for 1 hour. The mixture is then treated  
under ice-cooling with 3.6g NaBH<sub>4</sub> and stirred for 16 hours  
15 at room temperature. The resulting mixture is concentrated  
under vacuum, the residue partitioned between saturated  
NaHCO<sub>3</sub> and ethyl acetate and the organic phase dried and  
concentrated. The title substance is obtained by chromat-  
ography over kieselgel (eluant - toluene/ethyl acetate 4:1)  
20 as an oil.

The following compounds of formula I can be  
obtained in an analogous manner.



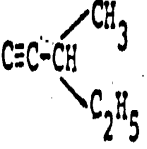
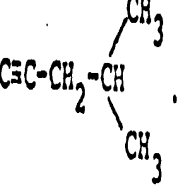

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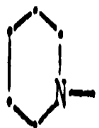
TABLE I


10260X



Example	R <sub>1</sub>	R <sub>2</sub>	R <sub>3</sub>	R <sub>4</sub>	R <sub>5</sub>	R <sub>6</sub>	Conf.	Physical data	Proc.
7		H	H	CH <sub>3</sub>	H	-C≡C-(CH <sub>2</sub> ) <sub>3</sub> -CH <sub>3</sub>	trans	oil	c, e
8	- " -	H	H	CH <sub>3</sub>	H	- " -	cis	oil	e
9		H	H	CH <sub>3</sub>	H	- " -	cis	oil	a, e
10		H	H	CH <sub>3</sub>	H	- " -	trans	oil	a, c, e
11	- " -	H	H	CH <sub>3</sub>	H	- " -	cis	oil	a, e
12		H	H	CH <sub>3</sub>	H	- " -	trans	oil	a, c, e
13	- " -	H	H	CH <sub>3</sub>	H	- " -	cis	oil	a, e
14		H	H	CH <sub>3</sub>	H	- C ≡ CH	trans	mp (hydrochloride) 150-155°	a, c, e


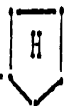







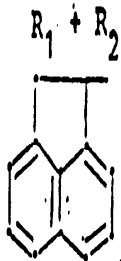

Example	R <sub>1</sub>	R <sub>2</sub>	R <sub>3</sub>	R <sub>4</sub>	R <sub>5</sub>	R <sub>6</sub>	Conf.	Physical data	Proc.
15		H	R <sub>3</sub> +R <sub>4</sub> +N 		H	-C≡CH	trans	mp (hydrochloride) 150-155°	a, c
16	- " -	H	H	CH <sub>3</sub>	H	-C≡C-C(CH <sub>3</sub> ) <sub>3</sub>	trans	m.p. (hydrochloride) 199-202° (crystal inversion above 135°)	a, c, e
17	- " -	H	H	CH <sub>3</sub>	H	- " -	cis	oil	a, e
18	- " -	H	H	CH <sub>3</sub>	H	-C≡C-C <sub>6</sub> H <sub>5</sub>	trans	oil	a, c, e
19	- " -	H	H	CH <sub>3</sub>	H	- " -	cis	oil	a, e
20	- " -	H	H	CH <sub>3</sub>	H		trans	m.p. (hydrochloride) 160-162°	a, c, e
21	- " -	H	H	CH <sub>3</sub>	H	- " -	cis	oil	a, e
22	- " -	H	H	CH <sub>3</sub>	H		trans	m.p. (hydrochloride) 124-126°	a, c, e
23	- " -	H	H	CH <sub>3</sub>	H	- " -	cis	oil	a, e
24	- " -	H	H	CH <sub>3</sub>	H		trans	oil	a, c, e

Example	R <sub>1</sub>	R <sub>2</sub>	R <sub>3</sub>	R <sub>4</sub>	R <sub>5</sub>	R <sub>6</sub>	Conf.	Physical data	Proc.
25		H	H	CH <sub>3</sub>	H	$\begin{array}{c} \text{OH} \\   \\ -\text{C}\equiv\text{C}-\text{C}-\text{CH}_3 \\   \\ \text{CH}_3 \end{array}$	trans	oil	c,e
26	- " -	H	H	CH <sub>3</sub>	H	$\begin{array}{c} \text{OH} \\   \\ -\text{C}\equiv\text{C}-\text{C}-\text{C}_2\text{H}_5 \\   \\ \text{C}_2\text{H}_5 \end{array}$	trans	oil	b,c,e
27	- " -	H	H	CH <sub>3</sub>	H	$\begin{array}{c} \text{OH} \\   \\ -\text{C}\equiv\text{C}-\text{CH}-(\text{CH}_2)_3-\text{CH}_3 \\   \\ \text{C}_2\text{H}_5 \end{array}$	trans	oil	b,c,e
28	- " -	H	H	CH <sub>3</sub>	H	$\begin{array}{c} \text{OH} \\   \\ -\text{C}\equiv\text{C}-\text{C}-\text{CH}_3 \\   \\ \text{C}_2\text{H}_5 \end{array}$	trans	oil	b,c,e
29	- " -	H	H	CH <sub>3</sub>	H	$\begin{array}{c} \text{OH} \\   \\ -\text{C}\equiv\text{C}-\text{C}-\text{CH}_3 \\   \\ \text{C}(\text{CH}_3)_3 \end{array}$	trans	oil	b,c,e
30	- " -	H	H	CH <sub>3</sub>	H	-C≡C-(CH <sub>2</sub> ) <sub>3</sub> -CH <sub>3</sub>	trans	mp (hydrochloride) 118-121°	a, e
31	- " -	H	H	CH <sub>3</sub>	H	-C≡C-(CH <sub>2</sub> ) <sub>2</sub> -CH <sub>3</sub>	trans	oil	a,c,e
32	- " -	H	H	CH <sub>3</sub>	H	-C≡C-(CH <sub>2</sub> ) <sub>4</sub> -CH <sub>3</sub>	trans	oil	a,c,c
33	- " -	H	H	CH <sub>3</sub>	H	-C≡C-(CH <sub>2</sub> ) <sub>5</sub> -CH <sub>3</sub>	trans	oil	a,c,e
34	- " -	H	R <sub>3</sub> +R <sub>4</sub> +N 		H	-C≡C-(CH <sub>2</sub> ) <sub>2</sub> -CH <sub>3</sub>	trans	oil	a,c
35	- " -	H	- " -		H	-C≡C-(CH <sub>2</sub> ) <sub>3</sub> -CH <sub>3</sub>	trans	oil	a,c

Example	R <sub>1</sub>	R <sub>2</sub>	R <sub>3</sub>	R <sub>4</sub>	R <sub>5</sub>	R <sub>6</sub>	Conf.	Physical data	Proc.
36		H	H	CH <sub>3</sub>	H	-C≡C-CH=CH-(CH <sub>2</sub> ) <sub>2</sub> -CH <sub>3</sub>	trans	oil	a,c,e
37	- " -	H	H	CH <sub>3</sub>	H	$  \begin{array}{c}  -C \equiv C - C = CH \cdot CH_3 \\    \\  C_2H_5  \end{array}  $	trans	oil	a,c,d,e
38	- " -	H	H	CH <sub>3</sub>	H	$  \begin{array}{c}  -C \equiv C - C = CH \cdot CH_3 \\    \\  CH_3  \end{array}  $	trans	oil	a,c,d,e
39	- " -	H	H	CH <sub>3</sub>	H	$  \begin{array}{c}  -C \equiv C - C = CH_2 \\    \\  C(CH_3)_3  \end{array}  $	trans	oil	a,c,d,e
40	- " -	H	H	CH <sub>3</sub>	H	$  \begin{array}{c}  -C = CH_2 \\    \\  C_6H_5  \end{array}  $	trans	oil	c,d,e
41	- " -	H	H	CH <sub>3</sub>	H	$  \begin{array}{c}  -C = CH_2 \\    \\  CH_2 \cdot \overset{\overset{CH_3}{ }}{CH} \\  \quad \quad \quad   \\  \quad \quad \quad CH_3  \end{array}  $	trans	oil	c,d,e
42	- " -	H	H	CH <sub>3</sub>	H	$  \begin{array}{c}  -C = CH_2 \\    \\  (CH_2)_3 \cdot CH_3  \end{array}  $	trans	oil	c,d,e
43	- " -	H	H	CH <sub>3</sub>	H	$  \begin{array}{c}  -C = CH_2 \\    \\  C(CH_3)_3  \end{array}  $	trans	oil	c,d,e

Example	R <sub>1</sub>	R <sub>2</sub>	R <sub>3</sub>	R <sub>4</sub>	R <sub>5</sub>	R <sub>6</sub>	Conf.	Physical data	Proc.
44		H	H	CH <sub>3</sub>	H	$\left\{ \begin{array}{l} \text{---} \overset{\text{CH}_3}{\underset{ }{\text{C}}} = \text{C}_6\text{H}_5 \\ \text{---} \overset{\text{CH}_2}{\underset{ }{\text{C}}} - \text{C}_6\text{H}_5 \end{array} \right.$	trans trans	oil oil	c,e c,e
45	- " -	H	H	CH <sub>3</sub>	H	- CH = C <sub>6</sub> H <sub>5</sub>	trans	oil	c,d
46	- " -	H	H	CH <sub>3</sub>	H	- C $\equiv$ C-CH <sub>2</sub> OH	trans	oil	b,c,e
47	- " -	H	H	CH <sub>3</sub>	CH <sub>3</sub>	- C $\equiv$ C-(CH <sub>2</sub> ) <sub>3</sub> -CH <sub>3</sub>	trans	oil	a,c,e
48	- " -	H	H	CH <sub>3</sub>	CH <sub>3</sub>	- " -	cis	oil	a,e
49	- " -	H	H	CH <sub>3</sub>	H	$\begin{array}{c} \text{CH}_3 \\   \\ \text{---} \text{C} \equiv \text{C} - \text{C} - \text{C}_2\text{H}_5 \\   \\ \text{CH}_3 \end{array}$	trans	oil	a,c,e
50	- " -	H	H	CH <sub>3</sub>	H	- " -	cis	oil	a,e
51	- " -	H	H	CH <sub>3</sub>	H	- C $\equiv$ C- 	trans	oil	a,c,e

Example	R <sub>1</sub>	R <sub>2</sub>	R <sub>3</sub>	R <sub>4</sub>	R <sub>5</sub>	R <sub>6</sub>	Conf.	Physical data	Proc.
52		H	H	CH <sub>3</sub>	H		cis	oil	a,e
53	- " -	H	H	CH <sub>3</sub>	H		trans	oil	a,c,e
54	- " -	H	H	CH <sub>3</sub>	H	- " -	cis	oil	a,e
55		H	H	CH <sub>3</sub>	H		trans	oil	c,d
56		H	H	CH <sub>3</sub>	H	-C≡C-C(CH <sub>3</sub> ) <sub>3</sub>	cis	oil	a,c,c

Example	R <sub>1</sub>	R <sub>2</sub>	R <sub>3</sub>	R <sub>4</sub>	R <sub>5</sub>	R <sub>6</sub>	Conf.	Physical data	Proc.
57		H	H	CH <sub>3</sub>	H	-C≡C-C(CH <sub>3</sub> ) <sub>3</sub>	cis	oil	a,e
58			H	CH <sub>3</sub>	H	-CH= 	trans	oil	c,d,e
59	- " -		H	CH <sub>3</sub>	H	-C≡C-C(CH <sub>3</sub> ) <sub>3</sub>	trans	oil	a,c,e

P In the following table NMR data are given. Data comprises peaks in ppm relative to TMS as standard in  $\text{CDCl}_3$ . Types of peaks are

P<sub>0</sub>

32

m = multiplet  
dt = double triplet  
dm = double multiplet  
s = singlet  
d = doublet  
t = triplet  
ps.t = pseudo triplet  
dd = double doublet  
dbr = double broad  
br = broad  
qua = quartet  
mbr = multiple broad  
sext = sextuplet  
ddd = double double doublet  
sbr = single broad

33

T0340X

Example	Isomer	Spectrum
1,7	trans	$\delta = 7.7-8.0$ (m, 2H); 7.15-7.45 (m, 4H); 6.14 (dt, $J=16$ and $2 \times 6.5$ Hz, 1 olef. H); 6.65 (dm, $J=16$ Hz, 1 olef. H); 3.72 (s, 2H); 3.10 (d, $J=6.5$ Hz, 2H); 2.3 (m, 2H); 2.24 (s, 3H); 1.2-1.7 (m, 4H); 0.9 (ps.t., 3H).
1,8	cis	$\delta = 7.7-8.0$ (m, 2H); 7.15-7.45 (m, 4H); 6.0 (dt, $J=11$ and $2 \times 6.5$ Hz, 1 olef. H); 5.64 (dm, $J=11$ Hz, 1 olef. H); 3.66 (s, 2H); 3.35 (d, $J=6.5$ Hz, 2H); 2.34 (m, 2H); 2.28 (s, 3H); 1.2-1.7 (m, 4H); 0.9 (ps.t., 3H).
9	cis	$\delta = 8.2-8.4$ (m, 1H); 7.7-7.9 (m, 2H); 7.3-7.6 (m, 4H); 6.05 (dt, $J=10.8 + 2 \times 7$ Hz, 1 olef. H); 5.65 (dm, $J=10.8$ Hz, 1 olef. H); 3.92 (s, 2H); 3.38 (dd, $J=7$ u. 1.5 Hz, 2H); 2.34 (m, 2H); 2.25 (s, 3H); 1.2-1.8 (m, 4H); 0.94 (m, 3H).
10	trans	$\delta = 6.9-7.2$ (m, 3H); 6.12 (dt, $J=16$ and $2 \times 6.5$ Hz, 1 olef. H); 5.64 (dm, $J=16$ Hz, 1 olef. H); 3.4 (s, 2H); 3.05 (d, $J=6.5$ Hz, 2H); 2.7-2.9 (m, 4H); 2.2-2.4 (m, 2H); 2.18 (s, 3H); 1.65-1.9 (m, 4H); 1.3-1.7 (m, 4H); 0.92 (m, 3H).
11	cis	$\delta = 6.85-7.2$ (m, 3H); 5.97 (dt, $J=11$ and $6.5$ Hz, 1 olef. H); 5.60 (dm, $J=11$ Hz, 1 olef. H); 3.45 (s, 2H); 3.30 (d, $J=6.5$ Hz, 2H); 2.7-2.9 (m, 4H); 2.2-2.4 (m, 2H); 2.22 (s, 3H); 1.7-1.9 (m, 4H); 1.3-1.7 (m, 4H); 0.95 (m, 3H).



Example	Isomer	Spectrum
12	trans	$\delta$ = 7.1-7.8 (m, 5H); 6.14 (dt, J=16 and 2 x 6.5 Hz, 1 olef. H); 5.65 (dm, J=16 Hz, 1 olef. H); 3.63 (s, 2H); 3.1 (d, J=6.5 Hz, 2H); 2.2-2.4 (m, 2H); 2.25 (s, 3H); 1.2-1.7 (m, 4H); 0.9 (m, 3H).
13	cis	$\delta$ = 7.1-7.8 (m, 5H); 6.0 (dt, J=11 and 2 x 6.5 Hz, 1 olef. H); 5.64 (dm, J=11 Hz, 1 olef. H); 3.66 (s, 2H); 3.35 (d, J=6.5 Hz, 2H); 2.2-2.4 (m, 2H); 2.30 (s, 3H); 1.2-1.7 (m, 4H); 0.9 (m, 3H).
16	trans	$\delta$ = 8.2-8.35 (m, 1H); 7.7-7.9 (m, 2H); 7.3-7.6 (m, 4H); 6.18 (dt, J=17 and 2x7 Hz); 5.65 (dm, J=17 Hz, 1H); 3.9 (s, 2H); 3.12 (dd, J=7 u. 1 Hz, 2H); 2.22 (s, 3H); 1.25 (s, 9H).
17	cis	$\delta$ = 8.2-8.35 (m, 1H); 7.7-7.9 (m, 2H); 7.3-7.6 (m, 4H); 6.03 (dt, J=11 and 2 x 6.5 Hz, 1H); 5.65 (dbr, J=11 Hz, 1H); 3.92 (s, 2H); 3.38 (d, J=6.5 Hz, 2H); 2.26 (s, 3H); 1.27 (s, 9H).
18	trans	$\delta$ = 8.2-8.35 (m, 1H); 7.7-7.9 (m, 2H); 7.2-7.6 (m, 9H); 6.36 (dt, J=16 and 2 x 6.5 Hz, 1H); 5.9 (dm, J=16 Hz, 1H); 3.94 (s, 2H); 3.22 (d, J=6.5 Hz, 2H); 2.28 (s, 3H).
19	cis	$\delta$ = 8.2-8.4 (m, 1H); 7.7-7.9 (m, 2H); 7.2-7.6 (m, 9H); 6.20 (dt, J=11 and 2 x 6.5 Hz, 1H); 5.85 (d, J=11 Hz, 1H); 3.98 (s, 2H); 3.50 (d, J=6.5 Hz, 2H); 2.30 (s, 3H).

35

Example	Isomer	Spectrum
20	trans	$\delta$ = 8.2-8.4 (m, 1H); 7.7-7.9 (m, 2H); 7.3-7.6 (m, 4H); 6.20 (dt, J=16 and 2 x 6.5 Hz, 1H); 5.80 (dm, J=16 Hz, 1H); 3.90 (s, 2H); 3.14 (d, J=6.5 Hz, 2H); 2.5 (m, 1H); 2.24 (s, 3H); 1.2-1.7 (m, 2H); 1.18 (d, J=7 Hz, 3H); 1.0 (t, J=7 Hz, 3H).
21	cis	$\delta$ = 8.2-8.4 (m, 1H); 7.7-7.9 (m, 2H); 7.3-7.6 (m, 4H); 6.05 (dt, J=11 and 2 x 6.5 Hz, 1H); 5.67 (dm, J=11 Hz, 1H); 3.94 (s, 2H); 3.40 (d, J=6.5 Hz, 2H); 2.55 (m, 1H); 2.28 (s, 3H); 1.2-1.8 (m, 2H); 1.20 (d, J=7 Hz, 3H); 1.02 (t, J=7 Hz, 3H).
22	trans	$\delta$ = 8.2-8.35 (m, 1H); 7.65-7.9 (m, 2H); 7.3-7.6 (m, 4H); 6.20 (dt, J=16 and 2 x 6.5 Hz, 1H); 5.68 (dm, J=16 Hz, 1H); 3.88 (s, 2H); 3.13 (d, J=6.5 Hz, 2H); 2.22 (s, 3H); 2.2 (m, 2H); 1.6-2.1 (m, 1H); 1.0 (d, J=7 Hz, 6H).
23	cis	$\delta$ = 8.2-8.4 (m, 1H); 7.7-7.9 (m, 2H); 7.3-7.6 (m, 4H); 6.04 (dt, J=12 and 2 x 7 Hz, 1H); 5.65 (dbr, J=12 Hz, 1H); 3.90 (s, 2H); 3.38 (d, J=7 Hz, 2H); 2.24 (s, 3H); 2.2 (m, 2H); 1.6-2.0 (m, 1H); 1.0 (d, J=7 Hz, 6H).
24	trans	$\delta$ = 8.2-8.4 (m, 1H); 7.65-7.9 (m, 2H); 7.3-7.6 (m, 4H); 7.15-7.3 (m, 2H); 6.95 (m, 1H); 6.36 (dt, J=16 u. 2 x 6 Hz, 1H); 5.9 (dbr, J=16 Hz, 1H); 3.92 (s, 2H); 3.20 (d, J=6 Hz, 2H); 2.28 (s, 3H).

36

Example	Isomer	Spectrum
2,25	trans	$\delta$ = 8.15-8.35 (m, 1H); 7.6-7.9 (m, 2H); 7.3-7.6 (m, 4H); 6.22 (dt, J=16 and 2 x 6.5 Hz, 1H); 5.67 (dt, J=16 and 2 x 1.5 Hz, 1H); 3.88 (s, 2H); 3.13 (dd, J=6.5 u. 1.5 Hz); 2.22 (s, 3H); 2.15 (br. -OH); 1.5 (s, 6H).
26	trans	identical with Ex. 2,25 except $\delta$ = 1.8 (br, OH); 1.65 (qua, J=8 Hz, 4H); 1.0 (t, J=8 Hz, 6H).
27	trans	$\delta$ = 8.2-8.35 (m, 1H); 7.6-7.9 (m, 2H); 7.3-7.6 (m, 4H); 6.26 (dt, J=16 and 2 x 6 Hz, 1H); 5.7 (dm, J=16 Hz, 1H); 4.46 (mbr, 1H); 3.90 (s, 2H); 3.15 (d, J=6 Hz, 2H); 2.25 (s, 3H); 2.0 (br, OH); 1.2-1.8 (m, 6H); 0.9 (m, 3H).
28	trans	$\delta$ = 8.15-8.35 (m, 1H); 7.7-7.9 (m, 2H); 7.3-7.6 (m, 4H); 6.25 (dt, J=16 and 2 x 6.5 Hz, 1 olef. H); 5.70 (dbr, J=16 Hz, 1H); 3.9 (s, 2H); 3.14 (d, J=6.5 Hz, 2H); 2.24 (s, 3H); 2.1 (br, OH); 1.72 (qua, J=7 Hz, 2H); 1.50 (s, 3H); 1.04 (t, J = 7 Hz, 3H).
29	trans	$\delta$ = 8.15-8.35 (m, 1H); 7.7-7.9 (m, 2H); 7.3-7.6 (m, 4H); 6.22 (dt, J=16 and 2 x 6.5 Hz, 1H); 5.70 (dm, J=16 Hz, 1H); 3.9 (s, 2H); 3.14 (d, J=6.5 Hz, 2H); 2.24 (s, 3H); 1.95 (m, OH); 1.46 (s, 3H); 1.06 (s, 9H).

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Example	Isomer	Spectrum
3,30	trans	$\delta$ = 8.2-8.35 (1 arom. H); 7.7-7.9 (2 arom. H); 7.3-7.6 (4 arom. H); 6.17 (dt, 1 olef. H, $J=16 + 2 \times 6.5$ Hz); 5.67 (d, 1 olef. H, $J=16$ Hz); 3.89 (s, 2H); 3.13 (d, 2H, $J=6.5$ Hz); 2.21 (s, 3H); 2.2-2.4 (m, 2H); 1.2-1.8 (4H); 0.8-1.05 (m, 3H).
31	trans	identical with Ex. 3,30 except: $\delta$ = 2.28 (t, 2H); 1.55 (sext., 2H); 1.0 (t, 3H).
32	trans	identical with Ex. 3,30 except: $\delta$ = 1.2 - 1.8 (m, 6H).
33	trans	identical with Ex. 3,30 except: $\delta$ = 1.2 - 1.8 (m, 8H).
34	trans	$\delta$ = 8.5 (br, 1H); 7.3-7.9 (m, 6H); 6.02 (ddd, $J=5, 8 + 16$ Hz, 1H); 5.46 (dbr, $J=16$ Hz, 1H); 3.80 (br, 1H); 3.1-3.35 (m, 2H); 2.52 (dd, $8 + 14$ Hz, 1H); 2.0-2.35 (m, 3H); 1.6-2.0 (m, 6H); 1.54 (sext., $J=7$ Hz, 2H); 0.97 (t, $J=7$ Hz, 3H).
35	trans	identical with Ex. 34 except: $\delta$ = 1.3-1.7 (m, 4H); 0.9 (ps.t, 3H).
4,36	trans	$\delta$ = 8.2-8.35 (m, 1H); 7.7-7.9 (m, 2H); 7.3-7.6 (m, 4H); 6.26 (dt, $J=15.5 + 2 \times 6.5$ Hz, 1H); 5.9 (dt, $J=11 + 2 \times 7$ Hz); 5.85 (d, $J=15.5$ Hz, 1H); 5.58 (dbr, $J=11$ Hz); 3.92 (s, 2H); 3.18 (d, $J=6.5$ Hz, 2H); 2.35 (t, 2H); 2.26 (s, 3H); 1.2-1.7 (m, 2H); 0.95 (ps.t. 3H).

38

Example	Isomer	Spectrum
37	trans	$\delta$ = 8.15-8.35 (m, 1H); 7.7-7.9 (m, 2H); 7.3-7.6 (m, 4H); 6.25 (dt, J=16 + 6 Hz, 1H); 5.86 (d, J=16 Hz, 1H); 5.70 (t, J=7 Hz, 1H); 3.94 (s, 2H); 3.20 (d, J=6 Hz, 2H); 2.26 (s, 3H); 2.16 (qua, J=8 Hz, 2H); [1.8 (d, J=7 Hz) und 1.7 (d, J=7 Hz); $\Sigma$ 3H, in ratio 6/1]; 1.06 (t, 3H).
38	trans	$\delta$ = 8.2-8.35 (m, 1H); 7.7-7.9 (m, 2H); 7.3-7.6 (m, 4H); 6.30 (dt, J=16 + 2x6 Hz, 1H); 5.86 (d, J=16 Hz, 1H); 5.75 (m, 1H); 3.92 (s, 2H); 3.18 (d, J=6 Hz, 2H); 2.26 (s, 3H); 1.87 (s, 3H); 1.8 u. 1.7 (2 d, 3H).
39	trans	$\delta$ = 8.2-8.4 (m, 1H); 7.7-7.9 (m, 2H); 7.3-7.6 (m, 4H); 6.28 (dt, J=16 + 2x6.5 Hz, 1H); 5.84 (dm, J=16 Hz, 1H); 5.30 (m, $\text{=C} \begin{smallmatrix} \text{H} \\ \text{H} \end{smallmatrix}$ ); 3.92 (s, 2H); 3.18 (d, J=6.5 Hz, 2H); 2.26 (s, 3H); 1.18 (s, 9H).
5,44 A	trans	$\delta$ = 8.2-8.35 (1 arom. H); 7.7-7.9 (2 arom. H); 7.3-7.6 (4 arom. H); 6.22 (d, 1 olef. H, J=16 Hz); 5.93 (dt, 1 olef. H, J=16 + 2 x 6.5 Hz); 4.87 u. 4.83 ( $\text{=C} \begin{smallmatrix} \text{H} \\ \text{H} \end{smallmatrix}$ ); 3.90 (s, 2H); 3.19 (d, 2H, J=6.5 Hz); 2.25 (s, 3H); 1.0-2.4 (11 H, Cyclohexyl).
B	trans	$\delta$ = 8.2-8.35 (1 arom. H); 7.7-7.9 (2 arom. H); 7.3-7.6 (4 arom. H); 6.79 (d, 1 olef. H, J=16 Hz); 5.80 (dt, 1 olef. H, J=16 + 2 x 6.5 Hz); 3.92 (s, 2H); 3.24 (d, 2H, J=6.5 Hz); 2.2-2.5 (m, 4H); 2.26 (s, 3H); 1.88 (s, 3H), 1.58 (br, 6H).

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Example	Isomer	Spectrum
40	trans	$\delta$ = 8.15-8.30 (m, 1H); 7.7-7.9 (m, 2H); 7.3-7.6 (m, 9H); 6.51 (d, J=18 Hz, 1H); 5.82 (dt, J=18 + 2 x 7.5 Hz, 1H); [5.26 (sbr, 1H) + 5.14 (d, J=2 Hz, 1H) =C $\begin{smallmatrix} \text{H} \\ \text{H} \end{smallmatrix}$ ]; 3.88 (s, 2H); 3.20 (d, J=7.5 Hz, 2H); 2.22 (s, 3H).
41	trans	$\delta$ = 8.2-8.35 (m, 1H); 7.7-7.9 (m, 2H); 7.3-7.6 (m, 4H); 6.24 (d, J=16 Hz, 1 olef. H); 5.85 (dt, J=16 + 2 x 6.5 Hz, 1 olef. H); 4.95 (dd, J=11 + 2 Hz, 2 olef. H); 3.9 (s, 2H); 3.18 (d, J=6.5 Hz, 2H); 2.24 (s, 2H); 2.13 (d, J=6.5 Hz, 2H); 1.6-2.1 (m, 1H); 0.9 (d, J=6.5 Hz, 6H).
42	trans	$\delta$ = 8.2-8.35 (m, 1H); 7.65-7.9 (m, 2H); 7.3-7.6 (m, 4H); 6.26 (d, J=16 Hz, 1H); 5.86 (dt, J=16 + 2 x 6.5 Hz, 1H); 4.95 (s, =C $\begin{smallmatrix} \text{H} \\ \text{H} \end{smallmatrix}$ ); 3.90 (s, 2H); 3.18 (d, J=6.5 Hz, 2H); 2.24 (s, 3H); 2.15-2.35 (m, 2H); 1.1-1.7 (m, 4H); 0.9 (ps.t, 3H).
43	trans	$\delta$ = 8.2-8.35 (m, 1H); 7.7-7.9 (m, 2H); 7.3-7.6 (m, 4H); 6.30 (d, J=15.5 Hz, 1H); 6.02 (dt, J=15.5 Hz + 2 x 6.5 Hz, 1H); [5.07 (sbr, 1H) + 4.80 (d, J=2 Hz, 1H), =C $\begin{smallmatrix} \text{H} \\ \text{H} \end{smallmatrix}$ ]; 3.9 (s, 2H); 3.16 (d, 2H); 2.25 (s, 3H); 1.1 (s, 9H).
6,45	trans	$\delta$ = 8.2-8.35 (1 arom. H); 7.7-7.9 (2 arom. H); 7.3-7.6 (4 arom. H); 6.52 (dd, 1 olef. H, J=15 u. 10 Hz); 5.86 (d, 1 olef. H, J=10 Hz); 5.79 (dt, 1 olef. H, J= 15 + 2 x 6.5 Hz); 3.92 (s, 2H); 3.20 (d, J= 6.5 Hz, 2H); 2.25 (s, 3H); 2.1-2.4 (m, 4H); 1.6 (br, 6H).

40

Example	Isomer	Spectrum
46	trans	$\delta$ = 8.15-8.35 (m, 1H); 7.7-7.9 (m, 2H); 7.3-7.6 (m, 4H); 6.3 (dt, $J=16 + 2 \times 6.5$ Hz, 1H); 5.7 (dm, $J=16$ Hz, 1H); 4.34 (d, $J=2$ Hz, 2H); 3.9 (s, 2H); 3.16 (d, $J=6.5$ Hz, 2H); 2.24 (s, 3H); 2.2 (CH).
47	trans	$\delta$ = 8.2-8.35 (m, 1H); 7.65-7.9 (m, 2H); 7.3-7.5 (m, 4H); 6.17 (dd, $J=16 + 7$ Hz, 1H); 5.58 (dm, $J=16$ Hz, 1H); 3.9 (AB-System, 2H); 3.25 (m, 1H); 2.1-2.3 (m, 2H); 2.14 (s, 3H); 1.3-1.6 (m, 4H); 1.18 (d, $J=7$ Hz, 3H); 0.85 (m, 3H).
48	cis	$\delta$ = 8.2-8.35 (m, 1H); 7.6-7.9 (m, 2H); 7.3-7.6 (m, 4H); 5.98 (dd, $J=11 + 9$ Hz, 1H); 5.6 (dm, $J=11$ Hz, 1H); 3.96 (AB-System, 2H); 3.8 (m, 1H); 2.1-2.3 (m, 2H); 2.16 (s, 3H); 1.2-1.6 (m, 4H); 1.26 (d, $J=7$ Hz, 3H); 0.82 (m, 3H).
49	trans	$\delta$ = 8.15-8.35 (m, 1H); 7.6-7.9 (m, 2H); 7.3-7.6 (m, 4H); 6.14 (dt, $J=16 + 2 \times 6.5$ Hz, 1H); 5.66 (dm, $J=16$ Hz, 1H); 3.86 (s, 2H); 3.10 (d, $J=6.5$ Hz, 2H); 2.2 (s, 3H); 1.4 (qua, $J=7$ Hz, 2H); 1.15 (s, 6H); 0.9 (t, $J=7$ Hz, 3H).
50	cis	$\delta$ = 8.2-8.35 (m, 1H); 7.6-7.9 (m, 2H); 7.3-7.6 (m, 4H); 6.0 (dt, $J=11 + 2 \times 6.5$ Hz, 1H); 5.64 (dm, $J=11$ Hz, 1H); 3.9 (s, 2H); 3.35 (d, $J=6.5$ Hz, 2H); 2.22 (s, 3H); 1.45 (qua, $J=7$ Hz, 2H); 1.18 (s, 6H); 0.95 (t, $J=7$ Hz, 3H).
51	trans	$\delta$ = 8.15-8.35 (m, 1H); 7.6-7.9 (m, 2H); 7.3-7.6 (m, 4H); 6.16 (dt, $J=16 + 2 \times 6.5$ Hz, 1H); 5.66 (dm, $J=16$ Hz, 1H); 3.86 (s, 2H); 3.10 (d, $J=6.5$ Hz, 2H); 2.7 (br, 1H); 2.2 (s, 3H); 1.4-2.1 (m, 6H).

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Example	Isomer	Spectrum
52	cis	$\delta$ = 8.15-8.35 (m, 1H); 7.6-7.9 (m, 2H); 7.3-7.6 (m, 4H); 6.0 (dt, $J=11 + 2 \times 6.5$ Hz, 1H); 5.64 (dm, $J=11$ Hz, 1H); 3.9 (s, 2H); 3.36 (d, $J=6.5$ Hz, 2H); 2.75 (br, 1H); 2.22 (s, 3H); 1.4-2.1 (m, 8H).
55	trans	$\delta$ = 7.8-8.1 (m, 2H); 7.25-7.5 (m, 3H); 6.50 (dd, $J=17 + 12$ Hz, 1H); 5.85 (d, $J=12$ Hz, 1H); 5.74 (dt, $J=17$ u. $2 \times 7$ Hz, 1H); 3.77 (s, 2H); 3.14 (d, $J=7$ Hz, 2H); 2.0-2.4 (m, 4H); 2.25 (s, 3H); 1.55 (sbr, 6H).
56	trans	$\delta$ = 8.2-8.4 (m, 2H); 7.25-7.7 (m, 3H); 6.74 (d, $J=8$ Hz, 1H); 6.2 (dt, $J=18 + 2 \times 7$ Hz, 1H); 5.67 (dt, $J=18$ u. $2 \times 15$ Hz, 1H); 4.0 (s, 3H); 3.82 (s, 2H); 3.10 (dd, $J=7$ u. $1.5$ Hz); 2.2 (s, 3H); 1.24 (s, 9H).
57	cis	$\delta$ = 8.2-8.4 (m, 2H); 7.25-7.7 (m, 3H); 6.74 (d, $J=8$ Hz, 1H); 5.05 (dt, $J=12 + 2 \times 7.5$ Hz, 1H); 5.65 (dt, $J=12$ u. $2 \times 1.5$ Hz, 1H); 4.0 (s, 3H); 3.85 (s, 2H); 3.35 (dd, $J=7.5$ u. $1.5$ Hz, 2H); 2.24 (s, 3H); 1.26 (s, 9H).
58	trans	$\delta$ = 7.2-7.8 (m, 6H); 6.44 (dd, $J=17 + 12$ Hz, 1H); 5.80 (d, $J=12$ Hz, 1H); 5.66 (dt, $J=17 + 2 \times 7$ Hz, 1H); 5.0 (t, $J=6$ Hz, 1H); 3.33 (d, $J=6$ Hz, 2H); 3.14 (d, $J=7$ Hz, 2H); 2.0-2.4 (m, 4H); 2.12 (s, 3H); 1.5 (sbr, 6H).
59	trans	$\delta$ = 7.1-7.7 (m, 6H); 6.04 (dt, $J=16 + 2 \times 6.5$ Hz, 1H); 5.6 (dm, $J=16$ Hz, 1H); 4.9 (t, $J=6$ Hz, 1H); 3.22 (d, $J=6$ Hz, 2H); 3.0 (d, $J=6.5$ Hz, 2H); 2.1 (s, 3H); 1.18 (s, 9H).



Example	Isomer	Spectrum
53	trans	$\delta$ = 8.15-8.35 (m, 1H); 7.6-7.9 (m, 2H); 7.3-7.6 (m, 4H); 6.15 (dt, $J=16 + 2 \times 6.5$ Hz, 1H); 5.65 (dm, $J=16$ Hz, 1H); 3.85 (s, 2H); 3.10 (d, $J=6.5$ Hz, 2H); 2.2 (s, 3H); 1.8-2.1 (br, 9H); 1.6-1.8 (br, 6H).

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P The required starting materials can be obtained  
e.g. as follows.

OL 1. Compounds of formula IV

P (A) (3-Benzo<sup>8 9</sup>[b]thiophenemethyl)methylamine (for Ex. 1)

5 P 3-Chloromethylbenzo<sup>8 9</sup>[b]thiophene is dissolved in  
benzene, added dropwise to a ca. 10-fold excess of  
methylamine in ethanol at 0-5° and then stirred for 16  
hours at room temperature. The crude mixture is concen-  
trated under vacuum, the residue partitioned between  
10 methylenechloride and 1N NaOH and the organic phase dried  
and evaporated under vacuum. The purified product is  
obtained by vacuum distillation b.p. 90-94°/1,33 Pa.  
14 20

P (B) (3-Benzo<sup>8 9</sup>[b]furanmethyl)methylamine (for Ex. 12 and 13)

P Obtained analogously to Example A)  
15 b.p. 105-110°/5.3 Pa.  
14 20

P (C) 2-(1-Naphthyl)piperidine (for Ex. 15, 34 and 35)

P A Grignard complex is prepared by adding 43.4g  
of 1-bromonaphthalene in absolute ether dropwise to 5.1g  
of magnesium in 50 ml of absolute ether. The ether is  
20 removed from the reaction mixture and replaced by absolute  
benzene. 8g 6-Methoxy-2,3,4,5-tetrahydropyridine are added  
to the boiling reaction mixture. After a further 8 hours  
the mixture is cooled, treated with saturated aqueous  
ammoniumchloride solution and the reaction product removed  
25 from the organic phase by shaking with aqueous HCl-solution.

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After neutralisation and working up the 2-(1-naphthyl)⊖  
3,4,5,6-tetrahydropyridine is dissolved directly in  
methanol and reduced with NaBH<sub>4</sub>. After normal working  
up the product is converted with alcoholic HCl solution  
5 to its hydrochloride. M.p. 287-289°<sup>14 20</sup> (after intensive  
drying under high vacuum 328-329°).  
<sup>14 20</sup>

CL 2. Compounds of formula V

P(D) 1-Bromo-6,6-dimethyl-2-hepten-4-yne (for Ex. 16, 17,  
56, 57. and 59)

P(a) 6,6-Dimethyl-1-hepten-4-yn-3-ole: <sub>A</sub> ~

10 38 ml 3,3-Dimethyl-1-butyne are dissolved in  
abs. tetrahydrofuran and 172 ml of a 20% solution of  
n-butyl-lithium added dropwise under protective gas at a  
temperature of -20°. The reaction mixture is then cooled  
to -75°<sup>31 20</sup> and 19.3 g acrolein in 20 ml of tetrahydrofuran  
15 added dropwise. The mixture is warmed to room temperature,  
reacted with saturated aqueous NH<sub>4</sub>Cl and extracted a  
number of times with ether. The organic phase is dried,  
concentrated and the purified product obtained by vacuum  
distillation, b.p. 70-72°/1600 Pa.  
<sup>14 20</sup>

20 P(b) 1-Bromo-6,6-dimethyl-2-hepten-4-yne: <sub>A</sub> ~

50 ml 48% HBr and 10g PBr<sub>3</sub> are stirred at 40°  
until a homogenous mixture is obtained. An alcoholic sol-  
ution of 13.5g 6,6-dimethyl-1-hepten-4-yn-3-ole are added

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dropwise at  $10^{\circ}$  and stirred for  $1\frac{1}{2}$  hours at room temperature. The reaction mixture is poured onto ice and extracted a number of times with hexane. The organic phase is washed a number of times with aqueous NaCl, dried and concentrated. NMR-spectrography shows that the oily product comprises a 3:1 mixture of trans- and cis-1-bromo-6,6-dimethyl-2-hepten-4-yne and is taken directly for alkylation.

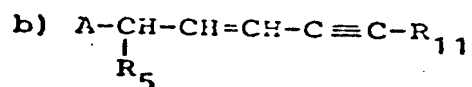
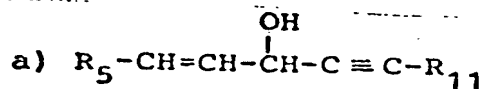
$\rho$  NMR:  $\delta = 5.5 - 6.4$  (m, 2 olef. H), [4.15 (d,  $J = 8\text{Hz}$ ) and 3.95 (d,  $J = 8\text{Hz}$ ) in ratio 1:3, 2H,  $\text{CH}-\text{CH}_2\text{Br}$ ], 1.20 (m, 9H).

$\rho$  Analogously to (D) above the following compounds of formula V can be obtained.

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T 0470X

Table II



	$R_{11}$	$R_5$	A	Physical data	for Ex.
E) a b	$\left\{ \begin{array}{l} \text{CH}_3 \\ -CH \\ \text{C}_2\text{H}_5 \end{array} \right.$	H	- Br	b.p. 75-80°/1460 Pa oil	20,21
F) a b	$\left\{ \begin{array}{l} \text{CH}_3 \\ -CH_2-CH \\ \text{CH}_3 \end{array} \right.$	H	- Br	b.p. 87-91°/1730 Pa oil	22,23
G) a b	$\left\{ \begin{array}{l} \text{CH}_3 \\ -C-C_2\text{H}_5 \\ \text{CH}_3 \end{array} \right.$	H	- Br	b.p. 90°/1460 Pa oil	49,50
H) a b	$\left\{ \begin{array}{l} \text{Cyclobutyl} \end{array} \right.$	H	- Br	b.p. 94-96°/800 Pa oil	51,52
I) a b	$\left\{ \begin{array}{l} -(CH_2)_3-CH_3 \end{array} \right.$	CH <sub>3</sub>	- Br	b.p. 92-93°/530 Pa oil	47,48

ρ The remaining compounds of formula V can be obtained analogously to (D) above.

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CL 3. Compounds of formula VIII

P (M) N-Methyl-N-(1-naphthylmethyl)octa-2,4-dienyl-1-amine  
(for Ex. 31)

P 9g 1,3-Heptadiyne, 16g methyl-(1-naphthylmethyl)⑦  
5 amine, 2.8g paraformaldehyde and 1.3g  $\text{ZnCl}_2$  (anhydrous)  
are heated for 3 hours at 100° in absolute dioxane. After  
cooling the solvent is removed <sup>20</sup> under vacuum, the residue  
partitioned between chloroform and aqueous  $\text{NaHCO}_3$ -solution  
and the organic phase dried and concentrated. The puri-  
10 fied product is obtained by chromatography over kieselgel  
(toluene/ethyl acetate 9:1) as an oil.

P (N) N-Methyl-N-(1-naphthylmethyl)-2,4-nonadienyl-1-amine  
(for Ex. 3)

P 8.25g 1-Bromohexyne are added dropwise to a mix-  
15 ture of 16g N-methyl-N-(1-naphthylmethyl)-propargylamine,  
0.5g  $\text{NH}_2\text{OH} \cdot \text{HCl}$ , 0.25g  $\text{CuCl}$  and 20 ml 70% ethylamine. The  
reaction mixture is stirred overnight at room temperature,  
treated with an aqueous solution of 1g  $\text{KCN}$  and extracted  
a number of times with ether. The organic phase is  
20 washed with saturated aqueous  $\text{NaCl}$ , dried and evaporated.  
The title substance is obtained as an oil after chromat-  
ography over Kieselgel (eluant toluene/ethyl acetate 95:5).

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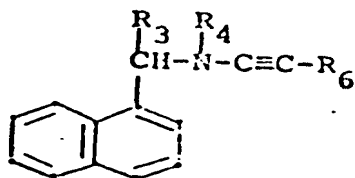
P (O) N-Methyl-N-(1-naphthylmethyl)-4-t-butyl-pent-2-yn-4-yl-1-amine (for Ex. 43)

P 933 mg N-Methyl-N-(1-naphthylmethyl)-4-hydroxy-4,5,5-trimethyl-2-hexynyl-1-amine are dissolved in abs. pyridine, warmed to 50° and 0.4 ml POCl<sub>3</sub> added. Stirring is carried out for one hour at 90°, the mixture poured onto ice and the reaction product isolated as an oil by extraction with ether and chromatography over kieselgel (eluant toluene/ethyl acetate 9:1).

Analogously to (M), (N) and (O) above, the following compounds of formula VIII may be obtained.

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Table III



	$\text{R}_3$	$\text{R}_4$	$\text{R}_6$	Physical data	For Ex.
P)	H	$\text{CH}_3$	$-\text{C}\equiv\text{C}-(\text{CH}_2)_4-\text{CH}_3$	oil	32
Q)	H	$\text{CH}_3$	$-\text{C}\equiv\text{C}-(\text{CH}_2)_5-\text{CH}_3$	oil	33
R)	H	$\text{CH}_3$	$-\text{C}\equiv\text{C}-\text{C}(\text{CH}_3)_3$	oil	16
S)	$\text{R}_3 + \text{R}_4 + \text{N}$		$-\text{C}\equiv\text{C}-(\text{CH}_2)_2-\text{CH}_3$	oil	34
T)			$-\text{C}\equiv\text{C}-(\text{CH}_2)_3-\text{CH}_3$	oil	35

*p* The remaining compounds of formula VIII can be prepared analogously to (M), (N) and (O) above.

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CL 4. Compounds of formula I

P (u) N-Methyl-N-(1-naphthylmethyl)-4-hydroxy-4-cyclohexyl-  
2-pentenyl-1-amine (for Ex. 5)

P (a) N-Methyl-N-(1-naphthylmethyl)-4-hydroxy-4-cyclohexyl-  
5 pent-2-enyl-1-amine  $\Delta$   $\sim$

P 10.7 ml of a 15% solution of BuLi in hexane are added dropwise to 3g N-methyl-N-(1-naphthylmethyl)propargyl amine in absolute tetrahydrofuran and after 30 minutes reacted with a solution of 1.79g cyclohexyl-  
10 methyl ketone. Stirring is continued for 24 hours at room temperature and the mixture poured onto ice and extracted with ether. The organic phase is washed, dried and concentrated under vacuum. Chromatography over kiesel-  
gel (eluant toluene/ethylacetate 4:1) yields the title  
15 product as an oil.

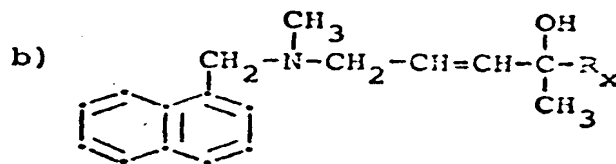
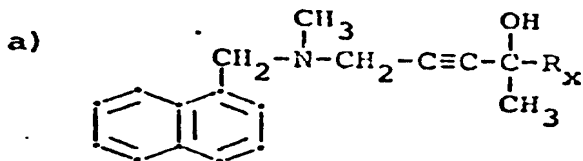
P (b) N-Methyl-N-(1-naphthylmethyl)-4-hydroxy-4-cyclohexyl-  
2-pentenyl-1-amine  $\Delta$   $\sim$

10g of the substance obtained under a) are dissolved in tetrahydrofuran and added dropwise to a sus-  
20 pension of 1.4g  $\text{LiAlH}_4$  in abs. tetrahydrofuran and the mixture refluxed for 3 hours. Excess reagent is destroyed with ethyl acetate/ $\text{H}_2\text{O}$ . After extraction with ether, drying and evaporation under vacuum followed by chromatog-  
raphy over kieselgel (eluant  $\text{CHCl}_3/\text{C}_2\text{H}_5\text{OH}$  95:5) the title  
25 product is obtained as an oil.

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† Analogously to (U) above the following compounds can be obtained.

Table IV



	R <sub>x</sub>	physical data [a) and b)]	For Ex.
v) a) b)	$\left. \begin{array}{l} \text{CH}_3 \\ \text{-CH}_2\text{-CH} \\ \text{CH}_3 \end{array} \right\}$	oil	41
w) a) b)	$\left. \begin{array}{l} \text{-(CH}_2\text{)}_3\text{-CH}_3 \end{array} \right\}$	oil	42
x) a) b)	$\left. \begin{array}{l} \text{-C(CH}_3\text{)}_3 \end{array} \right\}$	oil	43
y) a) b)	$\left. \begin{array}{l} \text{-C}_6\text{H}_5 \end{array} \right\}$	oil	40

Compounds of formula IX can be prepared analogously to Example 6 above and are preferably taken directly without further purification or isolation for the final step.

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T0530X

Example		Spectrum
N)		$\delta$ = 8.2-8.35 (1 arom. H); 7.7-7.9 (2 arom. H); 7.3-7.6 (4 arom. H); 3.97 (s, 2H); 3.37 (s, 2H); 2.40 (s, 3H); 2.2.-2.4 (m, 2H); 1.2-1.8 (4H); 0.8-1.05 (m, 3H).
M)		identical with N) except: $\delta$ = 2.28 (t, 2H); 1.58 (sext., 2H); 1.0 (t, 3H).
P)		identical with N) except: $\delta$ = 1.2-1.8 (m, 6H).
Q)		identical with N) except: $\delta$ = 1.2-1.8 (m, 8H).
R)		$\delta$ = 8.1-8.25 (m, 1H); 7.6-7.85 (m, 2H); 7.2-7.5 (m, 4H); 3.92 (s, 2H); 3.33 (s, 2H); 2.35 (s, 3H); 1.22 (s, 9H).
S)		$\delta$ = 8.5 (br, 1H); 7.3-7.9 (m, 6H); 4.05 (br, 1H); 3.24 (s, 2H); 3.12 (m, 1H); 2.5-2.8 (m, 1H); 2.26 (t, J=6.5 Hz, 2H); 1.6-2.0 (m, 6H); 1.56 (sext., J=7 Hz, 2H); 0.99 (t, J=7 Hz, 3H).
T)		identical with S) except: $\delta$ = 2.28 (ps.t, 2H); 1.3-1.7 (m, 4H); 0.91 (ps.t, 3H).

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Example		Spectrum
U)	a)	$\delta$ = 8.2-8.35 (1 arom. H); 7.7-7.9 (2 arom. H); 7.3-7.6 (4 arom. H); 4.0 (s, 2H); 3.37 (s, 2H); 2.38 (s, 3H); 1.52 (s, 3H); 1.0-2.2 (11H).
	b)	$\delta$ = 8.2-8.35 (1 arom. H); 7.7-7.9 (2 arom. H); 7.3-7.6 (4 arom. H); 5.76 (m, 2 olef. H); 3.91 (s, 2H); 3.13 (m, 2H); 2.25 (s, 3H); 1.23 (s, 3H); 0.8-2.0 (11H).
V)	a)	$\delta$ = 8.15-8.35 (m, 1H); 7.7-7.9 (m, 2H); 7.3-7.6 (m, 4H); 3.95 (s, 2H); 3.34 (s, 2H); 2.35 (s, 3H); 1.8-2.3 (m, 1H); 2.0 (s, OH); 1.62 (d, J=6.5 Hz, 2H); 1.53 (s, 3H); 1.04 u. 1.02 (2 d, J= 6.5 Hz, $\Sigma$ 6H).
	b)	$\delta$ = 8.2-8.4 (m, 1H); 7.7-7.9 (m, 2H); 7.3-7.6 (m, 4H); 5.78 (AB-portion of an ABX <sub>2</sub> -system, 2 olef. H); 3.90 (s, 2H); 3.12 (m, 2H); 2.22 (s, 3H); 1.3-2.0 (m, 1H); 1.5 (s, OH); 1.4 (d, 2H); 1.3 (s, 3H); 0.92 u. 0.90 (2 d, J=7 Hz, $\Sigma$ 6H).
W)	a)	$\delta$ = 8.2-8.35 (m, 1H); 7.7-7.9 (m, 2H); 7.3-7.6 (m, 4H); 3.98 (s, 2H); 3.36 (s, 2H); 2.38 (s, 3H); 2.1 (br, OH); 1.2-1.9 (m, 6H); 1.56 (s, 3H); 0.95 (ps.t., 3H).
	b)	$\delta$ = 8.2-8.35 (m, 1H); 7.7-7.9 (m, 2H); 7.3-7.6 (m, 4H); 5.85 (AB-portion of an ABX <sub>2</sub> -system, 2H); 3.90 (s, 2H); 3.12 (m, 2H); 2.25 (s, 3H); 1.2-1.7 (m, 6H + OH); 1.28 (s, 3H); 0.9 (ps.t., 3H).

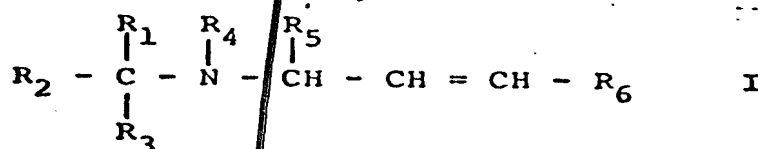
Example		Spectrum
X)	a)	$\delta = 8.2-8.35$ (m, 1H); 7.7-7.9 (m, 2H); 7.3-7.6 (m, 4H); 4.0 (s, 2H); 3.38 (s, 2H); 2.4 (s, 3H); 1.96 (br, OH); 1.54 (s, 3H); 1.14 (s, 9H).
	b)	$\delta = 8.2-8.4$ (m, 1H); 7.65-7.9 (m, 2H); 7.3-7.6 (m, 4H); 5.6-6.1 (AB-portion of an $ABX_2$ -system, $J=15 + 2 \times 5.5$ Hz, 2H); 3.92 (s, 2H); 3.16 (d, 2H; $J=5.5$ Hz); 2.25 (s, 3H); 1.4 (br, OH); 1.26 (s, 3H); 0.96 (s, 9H).
Y)	a)	$\delta = 8.2-8.35$ (m, 1H); 7.6-7.9 (m, 4H); 7.2-7.6 (m, 7H); 4.0 (s, 2H); 3.4 (s, 2H); 2.65 (br, OH); 2.4 (s, 3H); 1.85 (s, 3H).
	b)	8.15-8.35 (m, 1H); 7.65-7.9 (m, 2H); 7.2-7.6 (m, 9H); 5.6-6.1 (AB-portion of an $ABX_2$ -system, $J=15$ Hz + $2 \times 5.5$ Hz, 2H); 3.88 (s, 2H); 3.13 (d, $J=5.5$ Hz, 2H); 2.24 (s, 3H); 2.0 (s, OH); 1.65 (s, 3H).

CM I claim:

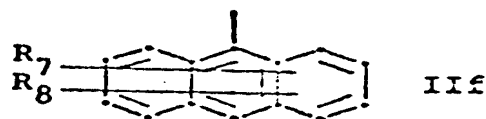
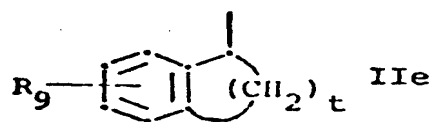
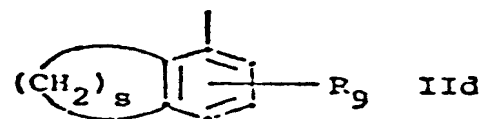
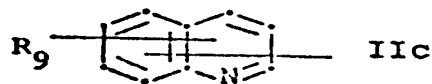
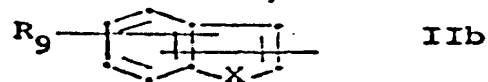
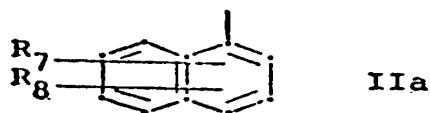
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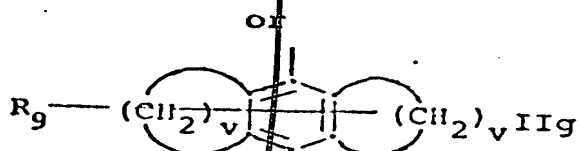
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1. A compound of formula I,

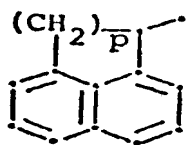


5 wherein a)  $R_1$  represents a group of formula

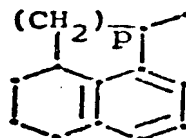




and  $R_2$  represents hydrogen or lower alkyl,  
or  $R_1$  and  $R_2$  together represent a group of formula



IIh



IIIi

whereby in the formulae IIa to IIIi,

$R_7$  and  $R_8$  represent, independently, hydrogen, halogen, tri-  
5 fluoromethyl, hydroxy, nitro, lower alkyl or lower alkoxy,  
 $R_9$  represents hydrogen, halogen, hydroxy, lower alkyl or  
lower alkoxy,

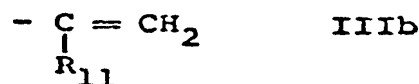
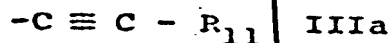
$X$  represents oxygen, sulphur, imino, lower alkyl imino or  
a radical of formula  $-(\text{CH}_2)_r-$ ,

10  $p$  is 1, 2 or 3,  
 $r$  is 1, 2 or 3,  
 $s$  is 3, 4 or 5,  
 $t$  is 2, 3 or 4, and  
 $v$  is 3, 4, 5 or 6;

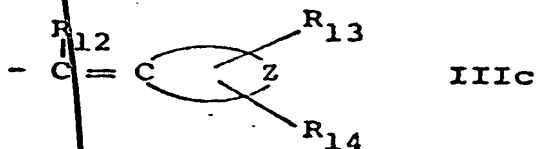
15  $R_3$  and  $R_5$  represent, independently, hydrogen or lower alkyl,  
and

$R_4$  represents  $C_{1-6}$  alkyl or  $C_{3-8}$  cycloalkyl- $(C_{1-6})$ -alkyl;  
and

$R_6$  represents a group of formula



or



wherein  $R_{11}$  represents hydrogen, optionally  $\alpha$ -hydroxy substituted alkyl; alkenyl, alkynyl, cycloalkyl, cycloalkylalkyl, phenyl, phenalkyl or thienyl,

5  $R_{12}$ ,  $R_{13}$  and  $R_{14}$  represent, independently, hydrogen or lower alkyl, and

$=C \text{---} Z$  represents a  $C_{5-8}$  cycloalkylidene radical optionally containing a double bond; or

10 b)  $R_1$  represents a group of formula IIa to IIg as defined under a),

$R_2$  represents hydrogen or lower alkyl,

$R_3$  and  $R_4$  together form a group  $-(CH_2)_u-$ , wherein  $u$  is an integer of 1 to 8, and

15  $R_5$  and  $R_6$  have the meanings given under a), or an acid addition salt thereof.

~~2. A compound as claimed in Claim 1 wherein~~